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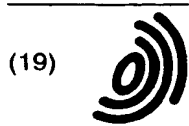
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(54) **CONTINUOUS MASS PRODUCTION OF CARBON NANOTUBES IN A NANO-AGGLOMERATE FLUIDIZED-BED AND THE REACTOR**

(57) The present invention relates to a method for continuous production of carbon nanotubes in a nano-agglomerate fluidized bed, which comprises the following steps: loading transition metal compounds on a support, obtaining supported nanosized metal catalysts by reducing or dissociating, catalytically decomposing a carbon-source gas, and growing carbon nanotubes on the catalyst support by chemical vapor deposition of carbon atoms. The carbon nanotubes are 4~100 nm in diameter and 0.5~1000  $\mu\text{m}$  in length. The carbon nanotube agglomerates, ranged between 1~1000  $\mu\text{m}$ , are smoothly fluidized under 0.005 to 2 m/s superficial gas velocity and 20~800  $\text{kg/m}^3$  bed density in the fluidized-bed reactor. The apparatus is simple and easy to operate, has a high reaction rate, and it can be used to produce carbon nanotubes with high degree of crystallization, high purity, and high yield.

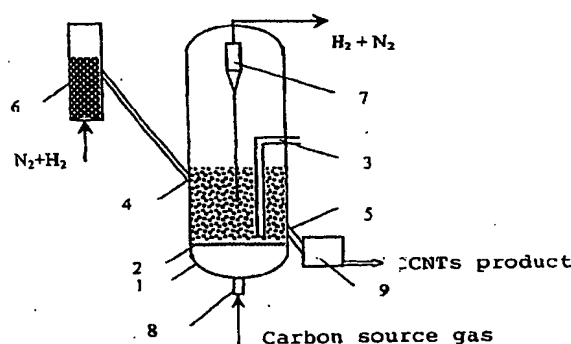


FIG. 1

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## Description

## FIELD OF THE INVENTION

[0001] The present invention relates to a method and reaction apparatus for continuous production of carbon nanotubes in a nano-agglomerate fluidized bed.

## BACKGROUND OF THE INVENTION

[0002] It is more than a decade since the first report on carbon nanotube as a new material. The exceptional mechanical and electrical properties of carbon nanotube have attracted intensive attention of physicists, chemists and material scientists worldwide, however, its commercial application has not been realized yet. The reasons lie in two interrelated aspects: the difficulty in mass production of carbon nanotubes and hence the high production cost. For instance, the international market price of carbon nanotubes of 90% purity is as high as \$60/g, which is 5 times that of gold. It is reported that the highest production rate of carbon nanotubes till now is only 200 g/h (MOTOO YUN4URA *et al.*, CNT10, Oct. 2001, p. 31). There are also reports forecasting that industrial application of carbon nanotubes will remain impractical until its price falls below \$2/pound, i.e. 0.4 cent/g, and it needs a production rate of 10,000,000 pound per year or about 12.5 tons per day to bring the price down to this level. Thus, in order to take carbon nanotubes from laboratory to market, mass production of high-quality carbon nanotubes is one of the principal challenges to take.

[0003] Novel process and reactor technology are the keys to the mass production of carbon nanotubes. Known methods for the preparation of carbon nanotubes mainly comprise graphite arc-discharge method, catalytic arc-evaporation method and catalytic decomposition method, of which the catalytic decomposition method is the most prevalent, especially the catalytic decomposition of lower hydrocarbons. The production of carbon nanotubes by chemical vapor deposition is a process involving both a typical chemical engineering process and the special preparation process of nanometer materials. Thus a desired production method should meet the requirements of heat transfer and mass transfer in the chemical engineering process while taking the special properties of nano-materials into consideration. Carbon nanotubes are one-dimensional nano-materials that grow during the reaction, and which demand a catalyst with its active ingredients dispersed on the nano-scale and which need sufficient space for growth. For a high rate of reaction, an appropriate concentration of catalysts is also necessary.

[0004] The gas-solid fluidization technique is an efficient measure to intensify the contact between gases and solids and has been widely used in many fields, and it is particularly suitable for the preparation, processing and utilization of powders. The gas-solid fluidization

technique offers many advantages, such as high throughput, large capacity of transporting/supplying heat, and easy transfer of powder products and catalysts. However, traditional gas-solid fluidized beds are only used for the fluidization of non-C-type powders with diameters larger than 30  $\mu\text{m}$  (Geldart D. *Powder Technology*, 1973, 7: 285). The growth of one dimensional materials and their adherence to each other in the preparation of carbon nanotubes by chemical vapor deposition tend to make fluidization difficult, and thus cause coagulation, uneven distribution of temperature and concentrations, and the deposition of carbon among particles. Therefore, there has been no report on the application of fluidized-bed reactor in continuous mass production of carbon nano-materials.

[0005] It is now known that the inter-particle forces among fine powders do not monotonically increase with the decrease of particle sizes. The intense Van der Waals force among nanometer particles can be effectively weakened in some nanomaterial systems by the formation of structurally loose agglomerates by the self-agglomeration of primary particles, which makes the said nano-materials fluidized and capable of flowing in the form of agglomerates. Chaouki *et al.* (*Powder Technology*, 1985, 43: 117) have reported the agglomerate fluidization of a  $\text{Cu}/\text{Al}_2\text{O}_3$  aerogel. Wang *et al.* (*Journal of Tsinghua University, Science and Technology Engineering*, vol.41, No.4/5, April 2001, p. 32-35) investigated the particulate fluidization behaviors of  $\text{SiO}_2$  nano-agglomerates. The agglomerate fluidization of carbon fibers was also reported by Brooks (*Fluidization V*, New York: Engineering Foundation, 1986, pp. 217).

## SUMMARY OF THE INVENTION

[0006] The object of the present invention is to provide a method and reaction apparatus for continuous production of carbon nanotubes in a nano-agglomerate fluidized bed, wherein the agglomeration and aggregation behaviors of nano-particles are taken into consideration. Normal fluidization state or even particulate fluidization state can be realized and maintained during the whole reaction process through proper control of the structure and growth of carbon nanotubes based on the analysis of the growth, agglomeration and fluidization of carbon nanotubes during the chemical vapor deposition process. By properly adjusting the reaction rate, operating conditions and fluidized-bed structure, the reactor bed is kept in an agglomerate fluidization state, so as to realize the continuous mass production of carbon nanotubes with a high degree of crystallization, high purity, and high yield.

[0007] The present invention provides a method for continuous production of carbon nanotubes in a nano-agglomerate fluidized bed, which comprises the following steps:

1. loading transition metal oxide on a support;

2. adding said transition metal oxide catalyst into a catalyst activation reactor, flowing a mixture of nitrogen and hydrogen or carbon monoxide into the reactor at 500~900°C to reduce the nanosized transition metal oxide particles to nanosized metal particles, wherein the volume ratio of hydrogen or carbon monoxide to nitrogen is from 1:0.3 to 1:1, the space velocity during the reduction reaction is from 0.3 h<sup>-1</sup> to 3h<sup>-1</sup> and the catalyst is in the form of nano-agglomerates, which have diameters between 1~1000 μm;

3: transporting the catalyst into a fluidized-bed reactor, flowing a mixture of hydrogen or carbon monoxide, a gas of lower hydrocarbons having less than 7 carbon atoms, and nitrogen into the reactor at 500~900°C, with the volume ratio of hydrogen or carbon monoxide:carbon-source-gas:nitrogen equals 0.4~1:1:0.1~2, wherein the space velocity during the reaction is 5~10000 h<sup>-1</sup>, the superficial gas velocity is 0.08~2 m/s, the bed density is maintained at 20~800 kg/m<sup>3</sup>, and the nano-agglomerates of the catalyst and the carbon nanotube product are kept in a dense-phase fluidization state, as a result, carbon nanotubes are obtained from the fluidized-bed reactor.

**[0008]** The process can be operated continuously when the catalyst and the reactants are fed continuously and the product is continuously removed out from the reactor.

**[0009]** According to the present invention, a second method for continuous production of carbon nanotubes in a nano-agglomerate fluidized bed comprises the following steps:

1. placing a catalyst support in the fluidized bed reactor, wherein the diameters of the agglomerates of the catalyst support are in the range of 1~1000 μm and the bed density of the reactor is 20~1500 kg/m<sup>3</sup> so that the catalyst support can be fluidized;
2. dissolving a metallocene compound in a low carbon number organic solvent;
3. heating the above solution to a temperature higher than the boiling point of the organic solvent to vaporize the solution;
4. feeding the above vaporized catalyst precursor into the fluidized-bed reactor, flowing a mixture of hydrogen or carbon monoxide, a gas of lower hydrocarbons having less than 7 carbon atoms, and nitrogen into the reactor at 500~900°C, with the volume ratio of hydrogen or carbon monoxide:carbon-source-gas:nitrogen equals 0.4~1:1:0.1~2, wherein the space velocity during the reaction is 5~10000 h<sup>-1</sup>, the superficial gas velocity is 0.005~2 m/s, and the stuffs in the reactor are kept in a dense-phase fluidization state, as a result, carbon nanotubes are obtained from the fluidized-bed reactor.

**[0010]** The present invention also provides a reaction apparatus for the continuous production of carbon nanotubes in a nano-agglomerate fluidized bed, which comprises a main reactor, a catalyst activation reactor, a gas distributor, a gas-solid separator and a product degassing section. The catalyst activation reactor is connected to the main reactor, the gas distributor is placed at the bottom of the main reactor, the gas-solid separator is arranged at the top of the main reactor, heat exchange tubes are provided inside the main reactor, and means for feeding gases are provided at the bottom of the main reactor, and the product degassing section is connected to the main reactor.

**[0011]** In the second method of the invention, due to the use of a metallocene compound as the catalyst precursor, the catalyst activation reactor can be omitted and the metallocene compound is directly fed into the main reactor containing the catalyst support, so that catalyst preparation and the main reaction are integrated.

**[0012]** According to the present invention, the catalyst and the carbon nanotube product, which exist in the form of agglomerates during the process, are kept in a state of good flowability/fluidization by control of the operation conditions.

**[0013]** The catalyst support can be selected from powders with good flowability, such as superfine glass beads, silicon dioxide, alumina and carbon nanotubes. By adopting the process, conditions and reactors of the present invention, carbon nanotubes having a loose agglomerated structure can be produced with agglomerate diameters of 1~1000 μm, bulk density of 20~800kg/m<sup>3</sup>, and with good flowability/fluidization properties.

**[0014]** The reaction apparatus of the present invention has the following outstanding characteristics:

1. It makes good use of the specific characteristics of the fluidized bed, and it has compact structure and good applicability.
2. The stuffs in the reactor are of an appropriate density such that they can be kept in a state of flow/fluidization, and this provides sufficient growing space for the carbon nanotubes and also obtains sufficient reaction capacity.
3. It can continuously supply the catalyst into and remove the carbon nanotube product out of the reactor, thus a continuous mass production can be achieved.
4. During the production of the nanosized carbon materials, the distribution of temperature and concentrations in the fluidized bed are uniform, and there is neither local overheating nor coagulation.
5. It can supply heat in and remove heat out of a scaled-up apparatus, and is suitable for the exothermic or endothermic catalytic decomposition processes.
6. The adaptability of the reactor system is excellent. The locations of the feed inlet and product outlet can be adjusted according to the requirements

of the reaction residence time and the structure of the products.

#### Example 1:

### BRIEF DESCRIPTION OF THE DRAWINGS

#### [0015]

Fig. 1 is a schematic diagram of the structure of the reaction apparatus of the present invention. In Fig. 1, 1 is the main reactor, 2 is the gas distributor, 3 is the heat exchanger, 4 is the inlet for the catalyst, 5 is the outlet for the product, 6 is the catalyst activation reactor, 7 is the gas-solid separator, 8 is the gas feed device, 9 is the product degassing section.

Fig. 2 is a Scanning Electron Microscope (SEM) photograph of the carbon nanotube agglomerate produced using the method and reaction apparatus of the present invention.

Fig. 3 is a Transmission Electron Microscope (TEM) photograph of the carbon nanotubes produced using the method and reaction apparatus of the present invention.

Fig. 4 is a High Resolution Transmission Electron Microscope (HRTEM) photograph of the carbon nanotubes produced using the method and reaction apparatus of the present invention.

### EMBODIMENTS OF THE INVENTION

[0016] As shown in Fig. 1, according to the present invention, the reaction apparatus for the continuous production of carbon nanotubes in a nano-agglomerate fluidized bed comprises a main reactor 1, a catalyst activation reactor 6, a gas distributor 2, a gas-solid separator 7 and a product degassing section 9. The catalyst activation reactor 6 is connected to the main reactor 1, the gas distributor 2 is placed in the bottom of the main reactor 1 and the gas-solid separator 7 is arranged at the top of the main reactor 1, the main reactor 1 is provided with heat exchange tubes 3 and means for feeding gases at its bottom, and the product degassing section 9 is connected to the main reactor 1 through a product outlet 5. The product outlet 5 can be used to adjust the amount of the stuffs in the main reactor. The product outlet 5 is connected to the product degassing section 9 for desorbing the organic materials absorbed on the product.

[0017] The contents of the present invention are described in details by the following examples. However, the examples are not intended to limit the scope of the invention.

#### [0018]

1. Loading Fe-Cu transition metal oxides on a SiO<sub>2</sub> support.

2. Adding the above supported catalyst into the catalyst activation reactor and carrying out the reduction reaction by flowing a mixture of hydrogen and nitrogen into the reactor at 650°C, wherein the volume ratio of hydrogen to nitrogen was 1:0.5 and the space velocity of the reduction reaction was 0.5 h<sup>-1</sup>.

3. Transporting the reduced catalyst into the fluidized bed with temperature at 700°C, feeding a mixture of hydrogen, ethylene and nitrogen into the reactor, wherein the volume ratio of H<sub>2</sub>:C<sub>2</sub>H<sub>4</sub>:N<sub>2</sub> was 1:1:1 and the space velocity during the reaction was kept at 10000 h<sup>-1</sup> and the superficial gas velocity was 0.5 m/s.

[0019] Fig. 2 shows a typical SEM photo of the carbon nanotubes produced in the example 1. The sample was directly obtained from the reactor and was not subjected to any purification or pulverization. The carbon nanotubes are in the form of agglomerates, and most of the agglomerates are near spherical in shape with diameters of less than 100 μm.

[0020] Fig. 3 shows a TEM photo of the above-mentioned sample. During sample preparation, a small quantity of the unpurified sample was dispersed in ethanol by ultrasonic wave, and then dripped onto a fine copper grid for Transmission Electron Microscopy observation. It can be seen from the figure that the carbon nanotubes are quite pure and have diameters of less than 10 nm, and the tubes are long and uniform in diameter.

[0021] Fig. 4 is a HRTEM photo of the sample, which was prepared by the same procedure as that for Fig. 3. From the figure, the carbon atom layers of the multi-wall carbon nanotube can be observed.

#### Example 2:

#### [0022]

1. Loading Ni-Cu transition metal oxides on a glass bead support.

2. Adding the above supported catalyst into the catalyst activation reactor and carrying out the reduction reaction by flowing a mixture of hydrogen and nitrogen into the reactor at 520°C, wherein the volume ratio of hydrogen to nitrogen was 1:1 and the space velocity of the reduction reaction was 2 h<sup>-1</sup>.

3. Transporting the reduced catalyst into the fluidized bed with temperature at 520°C, feeding a mixture of hydrogen, propylene and nitrogen into the reactor, wherein the volume ratio of H<sub>2</sub>:C<sub>3</sub>H<sub>6</sub>:N<sub>2</sub> is 1:1:1 and the space velocity during the reaction was

kept at  $5 \text{ h}^{-1}$  and the superficial gas velocity was  $0.09 \text{ m/s}$ .

#### Example 3:

[0023]

1. Loading Co-Mn transition metal oxides on a  $\text{Al}_2\text{O}_3$  support.
2. Adding the above supported catalyst into the catalyst activation reactor and carrying out the reduction reaction by flowing a mixture of hydrogen and nitrogen into the reactor at  $800^\circ\text{C}$ , wherein the volume ratio of hydrogen to nitrogen was 1:0.5 and the space velocity of the reduction reaction was  $0.3 \text{ h}^{-1}$ .
3. Transporting the reduced catalyst into the fluidized bed with temperature at  $870^\circ\text{C}$ , feeding a mixture of hydrogen, methane and nitrogen into the reactor, wherein the volume ratio of  $\text{H}_2:\text{CH}_4:\text{N}_2$  was 0.5:1:0.1 and the space velocity during the reaction was kept at  $5000 \text{ h}^{-1}$  and the superficial gas velocity was  $0.8 \text{ m/s}$ .

#### Example 4:

[0024]

1. Loading Ni transition metal oxide on a  $\text{Al}_2\text{O}_3$  support.
2. Adding the above supported catalyst into the catalyst activation reactor and carrying out the reduction reaction by flowing a mixture of carbon monoxide and nitrogen into the reactor at  $870^\circ\text{C}$ , wherein the volume ratio of carbon monoxide to nitrogen was 1:0.5 and the space velocity of the reduction reaction was  $3 \text{ h}^{-1}$ .
3. Transporting the reduced catalyst into the fluidized bed with temperature at  $870^\circ\text{C}$ , feeding a mixture of hydrogen, ethylene and nitrogen into the reactor, wherein the volume ratio of  $\text{H}_2:\text{C}_2\text{H}_4:\text{N}_2$  was 1:1:0.5 and the space velocity during the reaction was kept at  $8000 \text{ h}^{-1}$  and the superficial gas velocity was  $1.3 \text{ m/s}$ .

#### Example 5:

[0025]

1. Loading Ni-Cu transition metal oxides on a  $\text{Al}_2\text{O}_3$  support.
2. Adding the above supported catalyst into the catalyst activation reactor and carrying out the reduction reaction by flowing a mixture of hydrogen and nitrogen into the reactor at  $870^\circ\text{C}$ , wherein the volume ratio of hydrogen to nitrogen was 1:0.5 and the space velocity of the reduction reaction was  $0.5 \text{ h}^{-1}$ .
3. Transporting the reduced catalyst into the fluidized bed with temperature at  $870^\circ\text{C}$ , feeding a mix-

ture of hydrogen, methane and nitrogen into the reactor, wherein the volume ratio of  $\text{H}_2:\text{CH}_4:\text{N}_2$  was 1:1:0.5 and the space velocity during the reaction was kept at  $9000 \text{ h}^{-1}$  and the superficial gas velocity was  $1.7 \text{ m/s}$ .

#### Example 6:

[0026]

1. Carbon nanotubes were placed in the main reactor as catalyst support.
2. Dissolving ferrocene in benzene, vaporizing the solution, and then feeding the obtained vapor together with propylene and nitrogen into the main reactor at  $650^\circ\text{C}$ , wherein the volume ratio of propylene:nitrogen:benzene:ferrocene equals 1:0.3:0.2:0.02, the superficial gas velocity was  $0.1 \text{ m/s}$  and the space velocity was  $200 \text{ h}^{-1}$ , the ferrocene was dissociated to form metal nano-particles supported on the carbon nanotube supports, and under the catalytic action of the metal nano-particles, the carbon-source gas was decomposed and new carbon nanotubes were obtained.

#### Claims

1. A method for continuous production of carbon nanotubes in a nano-agglomerate fluidized bed, comprising the following steps:
  - 1) loading transition metal oxide on a support;
  - 2) adding said transition metal oxide catalyst into a catalyst activation reactor, flowing a mixture of nitrogen and hydrogen or carbon monoxide into the reactor at  $500\sim 900^\circ\text{C}$  to reduce the nanosized transition metal oxide particles to nanosized metal particles, wherein the volume ratio of hydrogen or carbon monoxide to nitrogen is from 1:0.3 to 1:1, the space velocity during the reduction reaction is from  $0.3 \text{ h}^{-1}$  to  $3 \text{ h}^{-1}$  and the catalyst is in the form of nano-agglomerates, which have diameters between  $1\sim 1000 \mu\text{m}$ ;
  - 3) transporting the catalyst into a fluidized-bed reactor, flowing a mixture of hydrogen or carbon monoxide, a gas of lower hydrocarbons having less than 7 carbon atoms, and nitrogen into the reactor at  $500\sim 900^\circ\text{C}$ , with the volume ratio of hydrogen or carbon monoxide:carbon-source-gas:nitrogen equals  $0.4\sim 1:0.1\sim 2$ , wherein the space velocity during the reaction is  $5\sim 10000 \text{ h}^{-1}$ , the superficial gas velocity is  $0.08\sim 2 \text{ m/s}$ , the bed density is maintained at  $20\sim 800 \text{ kg/m}^3$ , and the nano-agglomerates of the catalyst and the carbon nanotube product are kept in a dense-phase fluidization state, as a result, car-

bon nanotubes are obtained from the fluidized-bed reactor.

2. A method for continuous production of carbon nanotubes in a nano-agglomerate fluidized bed, comprising the following steps: 5

1) placing a catalyst support in the fluidized-bed reactor, wherein the diameters of the agglomerates of the catalyst support are in the range of 1~1000  $\mu\text{m}$  and the bed density of the reactor is 20~1500  $\text{kg/m}^3$  so that the catalyst support can be fluidized; 10

2) dissolving a metallocene compound in a low carbon number organic solvent; 15

3) heating the above solution to a temperature higher than the boiling point of the organic solvent to vaporize the solution;

4) feeding the above vaporized catalyst precursor into the fluidized-bed reactor, flowing a mixture of hydrogen or carbon monoxide, a gas of lower hydrocarbons having less than 7 carbon atoms, and nitrogen into the reactor at 20

500~900°C, with the volume ratio of hydrogen or carbon monoxide:carbon-source-gas:nitrogen equals 0.4~1:1:0.1~2, wherein the space velocity during the reaction is 5~10000  $\text{h}^{-1}$ , the superficial gas velocity is 0.005~2  $\text{m/s}$ , and the stuffs in the reactor are kept in a dense-phase fluidization state, as a result, carbon nanotubes 25

are obtained from the fluidized-bed reactor. 30

3. A reaction apparatus for the continuous production of carbon nanotubes in the nano-agglomerate fluidized bed, comprising a main reactor (1), a catalyst 35
- activation reactor (6), a gas distributor (2), a gas-solid separator (7) and a product degassing section (9), wherein the catalyst activation reactor (6) is connected to the main reactor (1), the gas distributor (2) is placed in the bottom of the main reactor 40
- (1), the gas-solid separator (7) is arranged at the top of the main reactor (1), the main reactor (1) is provided with heat exchange tubes (3) and means for feeding gases at its bottom, and the product degassing section (9) is connected to the main reactor 45
- (1) through a product outlet (5).

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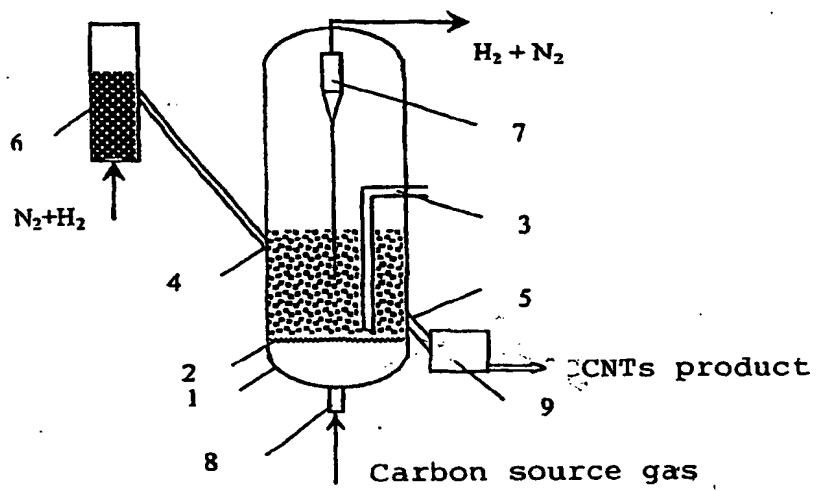


FIG. 1



FIG. 2

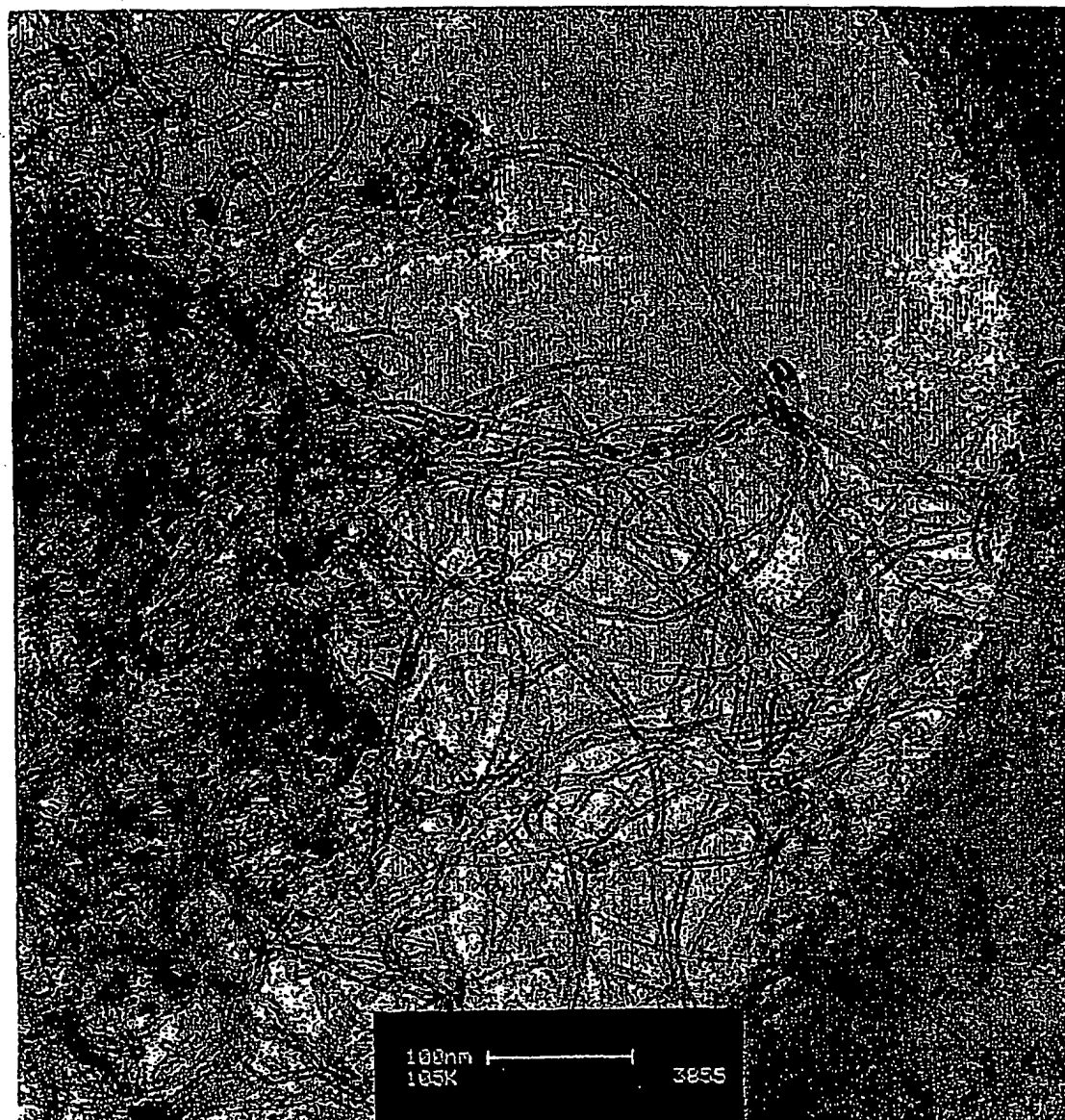


FIG. 3

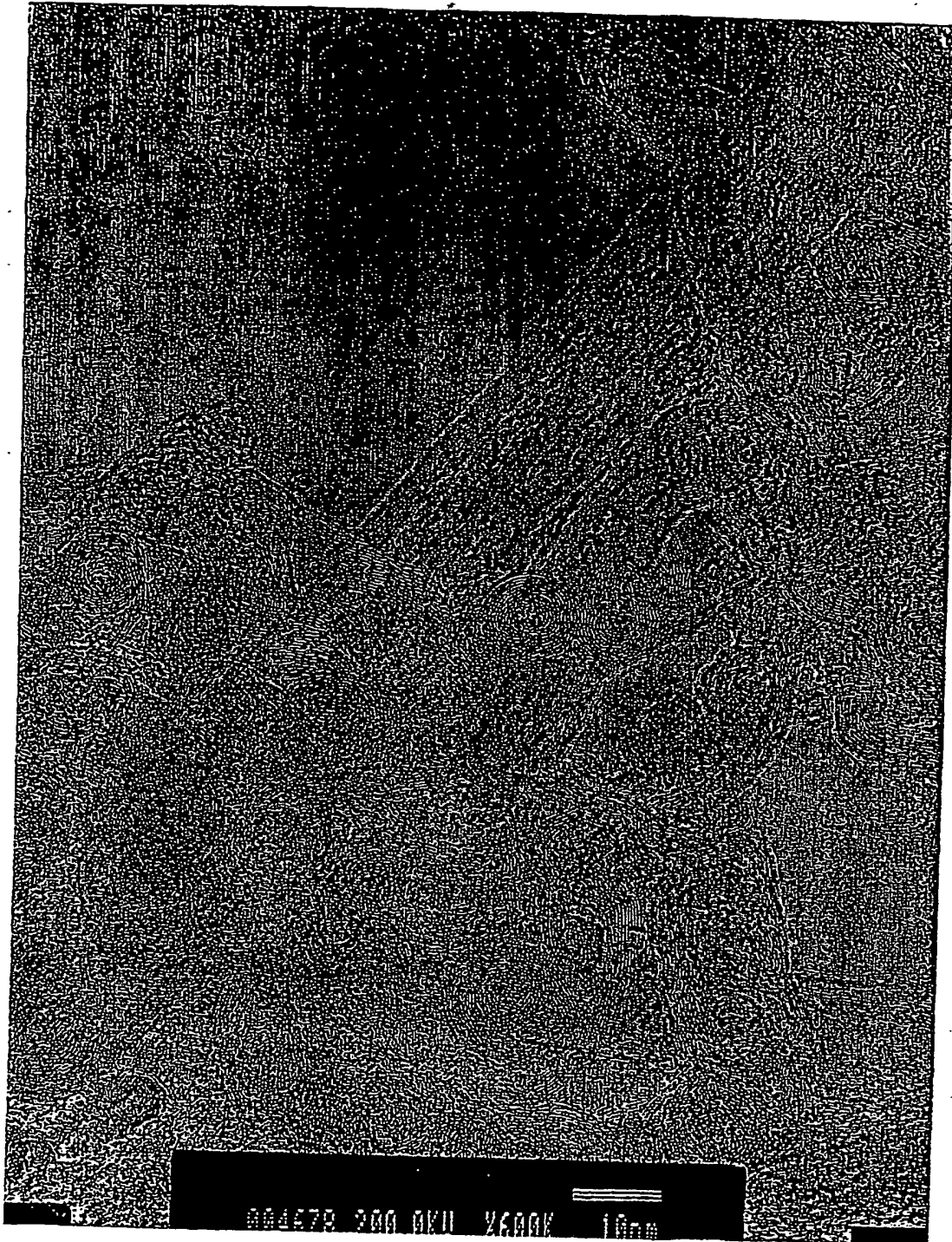


FIG. 4

## INTERNATIONAL SEARCH REPORT

International application No.

PCT/CN02/00044

## A. CLASSIFICATION OF SUBJECT MATTER

C01B31/02

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 C01B31/02

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Categor	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	CN1266018 ( UNIV TIANJIN ), 13.Sept.2000 , see the claim	1 , 2
A	CN1277145 ( LI TIEZHEN ), 20.Dec.2000 , see whole document	1 ~ 3
A	CN1315291 ( CHENGDU ORGANIC CHEM INS ), 3.Oct.2001 , see whole document	1 ~ 3
A	JP2000 - 203819 ( OSAKA GAS CO LTD ), 25.July.2000 , see whole document	1 ~ 3
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CN1315291	03.Oct.2001	none	
JP2000-203819	25.July.2000	none	
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